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Acid - contains hydrogen and produces  $H^+$   
+ ions when dissolved in water. Base -

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Answers  
contains hydroxide and produces  $\text{OH}^-$  ions when dissolved in water. Binary acids ( $\text{HF}$ ,  $\text{HCl}$ ,  $\text{H}_2\text{S}$ , etc.) Going across a row, rule is: the higher the electronegativity, stronger the acid. Ternary acids ( $\text{HNO}_3$ ,  $\text{H}_2\text{SO}_4$ ,  $\text{HClO}_4$ , etc.)

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Although the Arrhenius definitions of acid, base, and acid-base reactions provided in Sections 5.1, 5.4, and 5.6 are very important, especially to the beginning chemistry student, chemists have found it useful to extend these definitions to include new substances as acids and bases that would not be classified as such

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Answers according to the Arrhenius definitions.

*Chapter 5 Acids, Bases, and Acid-Base  
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Describe the characteristics of Arrhenius,

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acid. bases. Baking soda and antacid tablets are examples of two common \_\_\_\_\_ acid. a substance with a pH of 3 is a(n) bases. are often slippery substances that are used for cleaning solutions.

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1. What is

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## 16.1 Acids and Bases: A Brief Review.

- Acids taste sour and cause certain dyes to change color.
- Bases taste bitter and feel soapy.
- Arrhenius concept of acids and bases:
- An acid is a substance that, when

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dissolved in water, increases the concentration of  $H^+$  ions. •Example: HCl is an acid.

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Acids and Bases. An object's pH level can be tested using indicators. Objects with a

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low pH are acids, and those with a high pH are bases. Acids and bases react together to form water and salt.

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In this summary sheet study guide, you will find the key concepts of acids and

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Answers and acid-base reactions. It summarizes the factors such as electronegativity, polarizability, induction, and resonance-stabilization for determining the acidity of molecules and understating the  $pK_a$  values and relationship to acid strength.

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*Answersheet: Organic Acids and Bases  
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A strong acid is an acid which will dissociate completely to form an anion and a hydrogen ion. In contrast, a base is any type of molecule which can accept hydrogen ions from the solution. A ...



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Solutions are acidic or basic depending on their pH level, and this quiz/worksheet combo will test your understanding of the two concepts. Quiz & Worksheet Goals In these assessments, you'll be...

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*Quiz & Worksheet - Acids and Bases /  
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Study Guide: Acids and Bases 1) List at least three characteristic properties of acids and three of bases. ACIDS pH less than 7 Have a sour taste Change the color of many indicators Are corrosive (react

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Answers) Neutralize bases Conduct an electric current BASES pH greater than 7

## *KEY*

SAHOTA 04 Acid Base Study Guide-  
Written - Page 1 of 33. •Written Section:  
The questions below represent provincial  
exam questions(written) from Provincial

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Examinations in Chemistry dating back to Jan/94 to April 2000. It is important to do as many question types as possible. Once you have familiarized yourself with a particular type of problem and are confident in your abilities to solve it, signify it by placing checkmark on your.

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SAHOTA 03 Acid Base Study Guide -  
Multiple Choice - Page 1 of 47 Multiple  
Choice Section: This study guide is a  
compilation of questions from provincial  
exams since April 1994. I urge you to

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Answers become intimately familiar with question types. You will notice that questions from one year to another are very similar in their composition.

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What Are Acids and Bases? Lemons and



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Protons. When most of us hear the word acid, things such as lemons, batteries, and stomach juices come to... Arrhenius Gets a Make Over. Eventually, other chemists realized that some substances seemed to be acidic or basic but... Acid/Base Zombies vs. Acid/Base ...

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*What Are Acids and Bases? Help | Acids  
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An Acid is a type of sour substance. Examples of acids are lemon juice and vinegar. A base is a type of bitter substance. A base dissolved in water is called a basic solution.

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