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[6.2 6.2.1 Vinegar is a dilute form of acetic acid. A sample of acetic acid has the following percentage composition: 39,9% carbon 6,7% hydrogen 53,4% oxygen Determine the molecular formula of acetic acid if the molar mass of acetic acid is 60 g · mol⁻¹. \(6\) 6.2.2 Give a reason why acetic acid is regarded as a monoprotic acid. \(2\)](#)

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