

Solutions And Dilutions Pogil Home

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~~Solutions and Dilutions Shelby needs to make a 3.00 M acetic acid solution from an acetic acid standardized solution that is 15.0 M. The analytical equipment available to Shelby in the lab includes the volumetric flasks and pipets shown below.~~

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11/19: Molarity POGIL (both in-class), No HW 11/20: Molarity and Dilution Practice (answer keys in packet)- complete front AND back of last page for HW. CLICK HERE for video 11/21: Slushy Lab outside- wear warm clothing and bring gloves 11/22: Creating Solutions Lab- quiz grade for preparing solutions Week 2: 12/2: Acid/Base Cut and Sort ...

Unit 5 - MRS. FREEMAN'S CHEMISTRY SITE - Home

Dilution: Using solvent to increase the volume and thus decrease the solute concentration Some solutions call for solute amounts too small to weight out. Example: How much glucose would you need to make 50ml of a 1 uM solution (MW = 180g/mol)? Answer: 9 µg Make a concentrated stock solution then dilute it for use Dilutions

Laboratory Math II: Solutions and Dilutions

give a final solution concentration of 3 M: 20 mL into 100 mL; 10 mL into 50 mL; 5 mL into 25 mL. The dilution factor for all of these options is the same: 1/5. Diluting the acetic acid solution: stock 15 M desired 3 M pipette size flask size 5 10 15 20 25 500 0.15 0.3 0.45 0.6 0.75 250 0.3 0.6 0.9 1.2 1.5 100 0.75 1.5 2.25 3 3.75

Solutions to: Solutions and Dilutions

Thursday March 16 - Go over 14.1 and POGIL/ Finish 14.1 notes. Week of March 20. Monday March 20 - Notes: Solution Concentration. Homework: Solution Concentration. Tuesday March 21 - Notes: Dilutions/practice math. Start Notes: Solution Stoichiometry. Homework: Dilutions Worksheet. Wednesday March 22 - Finish Solution Stoichiometry

Chapter 14 - Honors | Ms. O'Connell's Homepage

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